

Design, Fabrication, and Inspection of Tanks for the Storage of Petroleum Refining Alkylation Unit Spent Sulfuric Acid at Ambient Temperatures

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Document History

2021-11-08: Reaffirmed by AMPP Standards Committee (SC) 16, Oil and Gas, Downstream.

2015-09-08: Reaffirmed by NACE TG (Task Group) 300,
“Petroleum Refining Spent Sulfuric Acid Storage Tank Requirements: Standard.”

2010-07-29: Reaffirmed by NACE TG 300,
“Petroleum Refining Spent Sulfuric Acid Storage Tank Requirements: Standard.”

2005-10-15: Approved as SP0205 by NACE TG 300,
“Petroleum Refining Spent Sulfuric Acid Storage Tank Requirements: Standard.”

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NACE SP0205-2021

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Foreword

Sulfuric acid (H_2SO_4) is the largest volume commodity chemical in use today. It plays some part in the production of nearly all manufactured goods. One large use of concentrated sulfuric acid is used as a catalyst for refinery alkylation units. In these units, C3-C5 olefins such as propylene or butylene are reacted with isobutane to form gasoline-blending components such as isoheptane and isooctane. These gasoline-blending components are used to boost octane for automobile and aviation fuels.

Most refineries have an alkylation unit that uses either hydrofluoric acid (HF) or sulfuric acid as the alkylation catalyst. This standard deals with spent sulfuric acid associated with the sulfuric acid alkylation process only.

Refineries using sulfuric acid alkylation typically require tanks for the storage of fresh (not yet used in the alkylation process) and spent (used in the alkylation process and in need of regeneration) acid. Design, fabrication, and inspection of fresh sulfuric acid tanks are covered in NACE SP0294.¹ This standard covers additions and deviations from SP0294 that apply to spent sulfuric acid storage tanks.

Large, vertical sulfuric acid storage tanks are usually built in accordance with API⁽¹⁾ Standard 650² or API Standard 620,³ and horizontal cylindrical tanks are built in accordance with the ASME⁽²⁾ Boiler and Pressure Vessel Code, Section VIII, Division 1.⁴ Although these codes and standards are sufficient for design strength and toughness considerations, they do not address the peculiarities of corrosion by alkylation unit spent sulfuric acid service. In addition, alkylation unit spent acid may contain dissolved hydrocarbons and hydrogen that release into the vapor space of these tanks and potentially produce an explosive environment. Thus, special care must be taken to deal with vapor leakage from the vapor space of these tanks and air intrusion into the vapor space of these tanks.

Carbon steel corrodes moderately when in contact with alkylation unit spent sulfuric acid. If tanks are properly designed and adequately maintained, use of carbon steel is an economical option for the storage of these acids at ambient temperatures. However, accelerated corrosion can occur in various forms, and a catastrophic failure of a spent acid tank in Delaware City, Delaware, U.S., has focused attention on the hazards associated with corrosion, vapor space leakage, and hot work on or around alkylation spent sulfuric acid tanks. The Chemical Safety Board⁽³⁾ report⁵ provides more details regarding this failure.

In AMPP standards, the terms *shall* and *must* are used to state requirements and are considered mandatory. The term *should* is used to state something that is recommended, but is not considered mandatory. The term *may* is used to state something considered optional.

⁽¹⁾ American Petroleum Institute (API), 1220 L St. NW, Washington, DC 20005-4070.

⁽²⁾ ASME, Two Park Avenue, New York, NY 10016-5990.

⁽³⁾ U.S. Chemical Safety and Hazard Investigation Board (CSB), 2175 K St. NW, Suite 400, Washington, DC 20037-1809.